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Key indicators

Single-crystal X-ray study T = 297 KMean $\sigma(\text{C}-\text{C}) = 0.002 \text{ Å}$ R factor = 0.045 wR factor = 0.128 Data-to-parameter ratio = 15.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. In the crystal structure of the title compound, $C_{15}H_{14}N_4O_3$, there are an intramolecular $O-H\cdots N$ hydrogen bond and intermolecular $\pi-\pi$ stacking interactions.

Comment

The title compound, (I) (Fig. 1), is a di-substituted pyrazole derivative. In this paper, we report the result of an X-ray diffraction study of this compound.



The pyrimidine, pyrazole and benzene rings are approximately coplanar; the dihedral angles between the pyrimidine and pyrazole, pyrimidine and benzene, and pyrazole and benzene rings are 3.50 (8), 11.84 (7) and 9.09 (8)°, respectively. The pyrazole rings at (x, y, z) and (2 - x, 2 - y, 1 - z), and the pyrimidine rings at (x, y, z) and (1 - x, 2 - y, 1 - z), form $\pi - \pi$ stacking interactions; the interplanar spacing and the centroid–centroid distance between the pyrazole rings are 3.464 (4) and 3.917 (2) Å, respectively, while those between the pyrimidine rings are 3.418 (2) and 3.810 (2) Å, respectively (Fig. 2).

Experimental

To a solution of 2-(1*H*-pyrazol-3-yl)-phenol (1.60 g, 10 mmol), which was prepared according to the reported method (Pleier *et al.*, 2001), and anhydrous K_2CO_3 (1.80 g, 13 mmol) in acetone (20 ml), 2-methanesulfonyl-4,6-dimethoxypyrimidine (2.18 g, 10 mmol) was added. The mixture was stirred at 329 K for 6 h. The resulting precipitate was purified by chromatography on silica gel with petroleum ether/acetone (15:1 ν/ν) as eluant to give (I) as a white solid (yield 2.47 g, 83%). Single crystals suitable for X-ray diffraction were obtained by crystallization from an ethanol solution at room temperature.

Crystal data

 $\begin{array}{l} C_{15}H_{14}N_4O_3\\ M_r = 298.30\\ \text{Monoclinic, } P2_1/c\\ a = 8.1971 \ (9) \ \text{\AA}\\ b = 11.1416 \ (12) \ \text{\AA}\\ c = 16.0917 \ (17) \ \text{\AA}\\ \beta = 98.538 \ (2)^{\circ}\\ V = 1453.3 \ (3) \ \text{\AA}^3 \end{array}$

Z = 4 $D_x = 1.363 \text{ Mg m}^{-3}$ Mo K α radiation $\mu = 0.10 \text{ mm}^{-1}$ T = 297 (2) K Block, colorless $0.40 \times 0.20 \times 0.04 \text{ mm}$

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Figure 1

The molecular structure of (I), showing the atom-labeling scheme. Displacement ellipsoids are drawn at the 50% probability level. The dashed line indicates the intramolecular hydrogen bond.

Data collection

Bruker SMART CCD area-detector	8620 measured reflections
diffractometer	3162 independent reflections
φ and ω scans	2470 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan	$R_{\rm int} = 0.022$
(SADABS; Sheldrick, 2001)	$\theta_{\rm max} = 27.0^{\circ}$
$T_{\min} = 0.962, \ T_{\max} = 0.996$	
Refinement	
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Refinement on F^2

 $w = 1/[\sigma^2(F_o^2) + (0.0729P)^2]$
$$\begin{split} R[F^2 > 2\sigma(F^2)] &= 0.045 \\ wR(F^2) &= 0.128 \end{split}$$
+ 0.1123P] where $P = (F_0^2 + 2F_c^2)/3$ S = 1.02 $(\Delta/\sigma)_{\rm max} = 0.001$ $\Delta \rho_{\rm max} = 0.14$ e Å 3162 reflections $\Delta \rho_{\rm min} = -0.24$ e Å⁻³ 204 parameters H atoms treated by a mixture of independent and constrained refinement

Table 1

Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$O1-H1\cdots N1$	0.92 (2)	1.78 (2)	2.6143 (16)	148 (2)

The O-bound H atom was located in a difference Fourier map and refined with a distance restraint of O-H = 0.92 (2) Å, and with $U_{iso}(H) = 1.5U_{eq}(O)$. All other H atoms were positioned geometrically (C-H = 0.93 or 0.96 Å) and refined as riding, with $U_{iso}(H) =$ $1.2U_{eq}(C)$ or $1.5U_{eq}(methyl C)$.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT-Plus (Bruker, 2001); data reduction: SAINT-Plus; program(s) used to



Figure 2



solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: PLATON.

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